OXIDATIVE DIMERIZATION OF N_b-METHOXYCARBONYLTRYPTAMINES BY DYE-SENSITIZED PHOTOOXYGENATION IN FORMIC ACID. SYNTHESIS OF $(^{\pm})$ -FolicAnthine and $(^{\pm})$ -chimonanthine

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Dyc-sensitized photooxygenation of N_b -methoxycarbonyltryptamines and tryptophan methyl ester in methanol at 0° has been disclosed to give 3a-hydroperoxypyrroloindoles (1) which were stable enough to be isolated and transformed into the hydroxy derivative (2), the ketoamide (3), and the N_b -formyl derivative (4) in various conditions such as silica-gel/methylene chloride, heat, or light.¹ Under similar conditions an aqueous solution of tryptophan afforded a tricyclic hydroperoxide (1) which gave the hydroxy derivative (2)in excellent yield on reduction and rearranged to formylkynurenine (3) on heating at 100°.²

On the other hand, proflavine-sensitized photooxygenation of tryptophan and N_b -benzyloxycarbonyltryptophan in formic acid has been reported to give kynurenine-type compounds in good yields.³ As the tricyclic hydroperoxide (1) rearranges to 1,4-benzoxazine derivative (5) in methanol-HCl at room temperature,¹ we reexamined proflavine-sensitized photooxygenation of tryptophan and N_b -benzyloxycarbonyltryptophan in formic acid, but neither kynureninetype compound nor 1,4-benzoxazine derivative was obtained, and polymeric compounds were isolated. Proflavine-sensitized photooxygenation of N_b -methoxy carbonyltryptamine in formic acid, however, was found to give dimeric products which have not previously been obtained in the sensitized photooxygenation of indole derivatives and were converted to folicanthine and chimonanthine.

When N_b -methoxycarbonyltryptamine (6) was irradiated in thoroughly 0_2 saturated formic acid at 10-15° with a 500W halogen lamp for 1 hr in the presence of proflavine hemisulfate (1/100 molar equivalent), N_a -formyl-3ahydroxypyrroloindole (7, 13-28%), mp 139-140°, and dimeric pyrroloindoles (8, 16-27%) as a mixture of two isomers were obtained.⁴ However, the 2,3-bond

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cleaved products were not isolated from this reaction. The structure of 7 was confirmed by the direct comparison with a sample obtained by the N_a-formylation of 2 (R₁= CO₂Me, R₂= H) and by the hydrolysis to 2 with MeOH-NaOH at room temperature. The dimeric compounds (8) could be separated by careful chromatography on silica-gel and recrystallization to 8a (racemic), mp 255-256°, and 8b (meso), mp 282-284°. 8a: $\lambda_{max}^{\text{EtOH}} \operatorname{nm}(\varepsilon)$; 247(21000),276^{sh} (4300), 286(3400). Mass $\underline{m}/\underline{e}(\%)$; 490(M⁺, 3), 462(M-CO, 7), 217((M-2CO)/2, 100). NMR(DMSO-d₆) δ ; 3.64(s, OMe), 6.06(s, NCHN), 8.88(s, CHO). 8b: $\lambda_{max}^{\text{EtOH}} \operatorname{nm}(\varepsilon)$; 249.5(20200), 276.5^{sh}(4300), 286.5(3200). Mass $\underline{m}/\underline{e}(\%)$; 490(M⁺, 16), 462(43), 217(100). NMR(DMSO-d₆) δ ; 3.34(s, OMe), 5.93(s, NCHN), 8.54 (br.s, CHO).

The dimers (8) were not formed when the reaction was carried out under N_2 -atmosphere or in acetic acid. The stereochemistry of 8 has been unequivocally established by LiAlH₄ reduction to chimonanthine and folicanthine which have been isolated from Calycanthaceous plants. Chimonanthine has been synthesized by Hendrickson⁵ and Scott⁶, while we have reported a synthesis of folicanthine.⁷ Scott and coworkers succeeded in the oxidative dimerization of N_b -methyltryptamine by ferric chloride oxidation of the indole Grignard reagent, but failed in ferricyanide oxidation of tryptamine derivatives.

LiAlH₄ reduction of a mixture of 8a and 8b in boiling ether for 20 hr afforded $(\stackrel{+}{-})$ -folicanthine, mp 167.5-168.5°, in 29% yield, which was identified with a sample obtained by our previous synthesis(mmp and IR). Its NMR spectrum in CDCl₃ was superimposable with that of natural folicanthine. On the other hand, LiAlH₄ reduction of 8b in boiling dioxane gave meso-folicanthine, mp 174-175°, hitherto not reported compound, in 60% yield. The UV and mass spectra of meso-folicanthine were very close to those of <u>rac</u>folicanthine, but differences were observed in their NMR spectra as shown in Table.

| Table | The | NMR | spectra | of | folicanthines | in | CDC1, |
|-------|-----|-----|---------|----|---------------|----|-------|
|-------|-----|-----|---------|----|---------------|----|-------|

| | с2, с3-н | N _b -Me | N _a -Me | NCHN | arom. H |
|---------|----------------------|--------------------|--------------------|-----------------------|----------------------|
| racemic | 1.6-2.8 ^m | 2.40 ^s | 3.00 ^s | 4.36 ^s | 6.1-7.1 ^m |
| meso | 1.8-3.0 ^m | 2. | 40 ^s | 3.6-4.6 ^{br} | 6.2-7.2 ^m |

Furthermore, hydrolysis of 8a with MeOH-10%NaOH to remove N_a-formyl group followed by LiAlH₄ reduction in boiling tetrahydrofuran provided (⁺)-chimonanthine, mp 184-186°, in 29% yield. Under similar conditions a mixture of 8a and 8b gave meso-chimonanthine, mp 198-203°, and (⁺)-chimonanthine. Spectral data(UV, mass, and NMR) and mp of these compounds agreed with those⁶ reported. Furthermore, (⁺)-chimonanthine was converted to (⁺)-calycanthine





in boiling acetic acid as reported. 6 Accordingly, 8a has racemic structure and 8b is meso isomer.

Similar results were obtained by the oxygenation of N_a -methyl- N_b -methoxycarbonyltryptamine and N_b -benzyloxycarbonyltryptamine. Both compounds gave the corresponding dimeric products which were converted to folicanthines. The mechanism of the oxidative dimerization by the sensitized photooxygenation is not clear at present, but simple hydrogen abstraction from N_a -H can be excluded due to the fact that the N_a -methyl derivative also provides dimeric products.

Although oxidative dimerization by dye-sensitized photooxygenation is restricted to N_b-acyltryptamines at present, it may represent an example for wider application of dye-sensitized photooxygenation in the biomimetic synthesis of natural products.

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References and notes

1. M. Nakagawa, H. Okajima, and T. Hino, <u>J. Am. Chem. Soc</u>., 99, 4424 (1977), 98, 635 (1976); M. Nakagawa, K. Yoshikawa, and T. Hino, <u>ibid</u>., 97, 6496 (1975), 96, 624 (1974); I.Saito, T. Matsuura, M. Nakagawa, and T. Hino, <u>Accounts</u> Chem. Res., 10, 346 (1977); M. Nakagawa, J. Chiba, and T. Hino, Heterocycles, 9, 385 (1978). 2. M. Nakagawa, H. Watanabe, S. Kodato, H. Okajima, T. Hino, J. L. Flippen, and B. Witkop, Proc. Natl. Acad. Sci. USA, 74, 4730 (1977). 3. C. A. Benassi, E. Scoffone, G. Galiazzo, and G. Iori, Phtochem. Photobiol., 6, 857 (1967). 4. Another dimeric product, $C_{26}H_{26}O_7N_4$, mp 247-248°, was also obtained with its isomer. Their structures will be discussed in a full paper. 5. J. B. Hendrickson, R. Goschke, and R. Rees, Proc. Chem. Soc., 1962, 383; Tetrahedron, 20, 565 (1964). 6. A. I. Scott, F. MacCapra, and E. S. Hall, <u>J. Am. Chem. Soc.</u>, <u>86</u>, 302 (1964); E. S. Hall, F. MacCapra, and A. I. Scott, <u>Tetrahedron,</u> 23, 4131 (1967). 7. T. Hino and S. Yamada, <u>Tetrahedron Letters</u>, 1963, 1757.

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